



Research Article

ISSN 2474-8811

**Photoluminescence Properties of Mn²⁺ Doped AlQ₃ Organic Phosphors for
Oled Devices and Flat Panel Displays****S.A. Bhagat**

Kamla Nehru College, Sakardara Square, Nagpur-440 024, India

Abstract

Pure and Mn²⁺ transition metal doped Alq₃ Complexes were synthesized by simple precipitation method at room temperature, maintaining stoichiometric ratio. These complexes were characterized by XRD, and photoluminescence (PL) spectra. XRD analysis exhibits the polycrystalline nature of the synthesized complexes. The excitation spectra are in the range of 416 nm to 438 nm. Prepared phosphors can be a suitable candidate for green and blue emitting OLED, PLLCD and solid state lighting applications.

Keywords: Transition metal, Metal complexes, Photoluminescence, OLED**Corresponding author:** S.A. Bhagat

Kamla Nehru College, Sakardara Square, Nagpur-440 024, India.
Tel: 9423064391,
E-mail: sunilbhagat15@rediffmail.com

Citation: S.A. Bhagat (2020), Photoluminescence Properties of Mn²⁺ Doped Alq₃ Organic Phosphors for Oled Devices and Flat Panel Displays. Int J Nano Med & Eng. 5:1

Copyright: ©2020 S.A. Bhagat. This is an open-access article distributed under the terms of the Creative Commons Attribution License.

Introduction

Now a day, organic light emitting materials are attracted attention owing to their applications in OLEDs^[1] for industry and academic research. It stimulates interest in next-generation displays and lighting technologies^[2]. The efficient and stable tri (8-hydroxyquinoline) aluminum (Alq₃) is extensively studied for its high stability, good emission, easy synthesis, electron transport properties and high quantum efficiency. Alq₃ is attributed as electron transporting layer, as emission layer where green light emission is generated by electron hole recombination in Alq₃. It also acts as host material for various dyes to tune emission color from red to green^[3]. Although Alq₃ has low fluorescence efficiency, it has excellent properties as emitting